Modification and Optimization of the Bis-picolylamide-Based Relay Protection for Carboxylic Acids to be Cleaved by Unusual Complexation with $Cu²⁺$ Salts

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S Supporting Information

[AB](#page-11-0)STRACT: [A simple mod](#page-11-0)ification of our recently published protection scheme for carboxylic acids as amides resulted in a new protecting group with significantly improved properties. It requires shorter reaction times for deprotection and allows us to replace $Cu(OTf)$ ₂ by $CuCl₂$, indicating at the same time the importance of the nature of the anion of the Cu^{2+} source. Since the new scheme fulfills all criteria required for an ideal protection group it should find widespread application in synthetic organic chemistry.

■ INTRODUCTION

The preparation of complex organic molecules in most cases encompasses the need for protecting groups.^{1,2} Common requirements for them are straightforward introduction and robustness to a variety of reaction conditions c[om](#page-11-0)bined with the possibility for selective and efficient cleavage, preferably under mild conditions. There is only a small number of protection schemes fulfilling all these criteria.

Recently, we reported on two new linker systems for solidphase chemistry which are based on chelating amides.^{3,4} These activities led us to the development of a relay protection scheme for carboxylic acids 1a as amides of bispic[oly](#page-11-0)lamine (bpa, 2) (Scheme 1).⁵ The protection group is easily removed from amide 3 by treatment with $Cu(OTf)_2$ in MeOH, which leads to quantitati[ve](#page-1-0) [fo](#page-11-0)rmation of methyl ester 4. The reaction is based on the unusual involvement of the amide nitrogen in the complexation to Cu^{2+} leading to a weakening of the amide bond which makes it accessible to the nucleophilic attack by MeOH.

Alternatively, the methanolysis was performed in the presence of $Ba(OH)_2·8H_2O^{6-8}$ which yielded after acidic workup the pertinent carboxylic acid 1b.

Compared to the commo[nly u](#page-11-0)sed protection of carboxylic acids as esters, the protection as amides shows a significantly higher stability due to the high resonance energy of about 80 kJ/mol. Despite its robustness it allowed for a cleavage under mild conditions. In addition, the cleavage process was orthogonal to most conditions employed for the deprotection of esters. Furthermore, bpa protection could be successfully applied to aliphatic, aromatic, and amino acids. A disadvantage though was the relatively long reaction time required for the methanolysis step which took between 16 and about 30 h and the use of the rather expensive $Cu(OTf)$ ₂ as copper source which had to be used in equimolar amounts.

■ RESULTS AND DISCUSSION

To widen the scope of the new protection scheme and to overcome the disadvantage of the long reaction times we carried out a systematic investigation of the influence of pendant coordinating groups on the rate of methanolysis of amides. Therefore, we prepared an array of variants to the original bpa entity (Scheme 2). The modifications involved the replacement of one or both pyridine rings of bis-picolylamine (bpa) by other coordinating [e](#page-2-0)ntities, as well as variations of the distances between the pyridyl donor sites.^{9,10}

Compounds 6−25 were then submitted to a cleavage process using $Cu(OTf)$ ₂ in MeOH at rt for 24 h[. Th](#page-11-0)e amount of the released benzoic acid methyl ester was quantified by HPLC using phenol as standard. As a benchmark for the methanolysis we used the transformation of the original bpa-protected benzoic acid amide 5 to its methyl ester (Table 1).

Omitting one pyridine ring and hence omitting one coordination site as in 6 led only to a marginal [c](#page-5-0)leavage rate, indicating that at least three coordinating sites are necessary to obtain decent methanolysis rates. Replacing one pyridine ring by other heterocycles was to no avail as well. Of all investigated compounds, derivatives 9, 10, and 25 revealed quantitative cleavage and were superior compared to the benchmark amide 5. Since the parent amine of derivative 9 is most easily obtained, very robust and structurally very simple, we focused on the application of amine 9a namely dimethylaminoethylpicolylamine (dmepa) for the protection of carboxylic acids.

For this reason, we performed a side by side evaluation of 9 with the bpa amide 5 using $Cu(OTf)$ ₂ as copper source. In the methanolysis of 9 at rt a quantitative conversion was already observed after about 5 h compared to the more than 30 h

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Scheme 1. Principle of the bpa-Based Protection for Carboxylic Acids As Chelating Amides

needed for derivative 5. Variations of the experimental conditions led to further enhancements. Addition of diisopropylethylamine (DIEA, Hü nig's base) turned out to have a positive effect on the cleavage rate which was also true for elevated temperatures. Hence, we also carried out the comparison at 50 °C and in the presence or absence of Hünig's base (Figure 1).

Quantitative cleavage of 9 was now possible within about 1 h.

Next we in[ve](#page-5-0)stigated the cleavage rate of an aliphatic amide. For this purpose, benzoylpropionic acid was first coupled to the parent dimethylaminoethylpicolylamine (dmepa) 9a using TBTU as coupling reagent (Scheme 3). The resulting amide 27 was then subjected to the methanolysis reactions and compared to the one of 26.

It turned out that formation of the e[xp](#page-6-0)ected methyl ester was at rt significantly slower than for the aromatic amide 9. Again, increasing the temperature to 50 °C and/or adding DIEA led to much faster cleavage of the amide and complete conversion was achieved within 2−3 h (Figure 2).

No differences were observed in the methanolysis rate of 28 and 29 (Scheme 4) applying $CuCl₂$ $CuCl₂$ $CuCl₂$ when compared to the one of 9. This indicated that electronic factors on the aromatic system had virtu[al](#page-6-0)ly no influence on this reaction.

A previous kinetic study involving bpa-amide 5 with either $Cu(OTf)_2$ and $CuCl_2$ had revealed that the reaction with the former proceeded much faster. For this reason, we had originally devised $Cu(OTf)_{2}$ as source for the Cu^{2+} being aware of the fact that this salt is rather expensive and that this would become an issue especially in large-scale synthesis employing equimolar amounts. To our delight, we discovered that the order of reactivity concerning the Cu source is reversed for the new derivative 9 (Figure 3). Complete transfer into the methyl ester was observed with $CuCl₂$ at rt already after about 1 h. This result emphasizes the i[mp](#page-6-0)ortance of the nature of the anion of the Cu^{2+} source, a topic of ongoing investigations. Application of $Ba(OH)_2·8H_2O$ in combination with the Cu^{2+} salt, furnished after acidic workup the carboxylic acid.

■ **CONCLUSIONS**

In summary, by investigating a number of structural variants of our original bpa protecting group for carboxylic acids we have developed a new and more versatile protecting group in which one pyridine entity of bpa was replaced by a simple tertiary amine and which we have dubbed as "dmepa" protecting group. The synthesis of the parent amine 9a was performed by a simple reductive amination between pyridine-2-carboxaldehyde and N,N-dimethylethylenediamine, both of which are commercially available. The methanolysis reaction of pertinent carboxylic acid amides was significantly faster compared to the one resulting from bpa and it could be further increased by running the reactions at 50 °C or in the presence of DIEA. Furthermore, the commonly used $Cu(OTf)_{2}$ could be replaced

by the much more effective and significantly cheaper $CuCl₂$. The new protection scheme works for aromatic as well as aliphatic carboxylic acids, and complete methanolysis can be achieved in both cases in less than 1 h. Thus, the newly developed protecting group for carboxylic acids fulfills virtually all requirements for an ideal protection scheme: The synthesis of the parent dmep-amine 9a comprises only one step starting from cheap starting materials. The protection of carboxylic acids is easily performed using TBTU as coupling reagent, 11 the protection itself is very robust, the rather fast cleavage is performed under mild conditions with high yields and [le](#page-11-0)ads optionally to the methyl ester or the free carboxylic acid, and the procedure for cleavage is orthogonal to most deprotection schemes for carboxylic acids. Hence, we believe that the newly developed protection scheme should find widespread application in synthetic organic chemistry.

EXPERIMENTAL SECTION

General Methods. All reactions with air- and moisture-sensitive compounds were carried out under an argon atmosphere. All solvents were analytical grade or better. Dry DCM and 1,2-DCE were distilled from $\rm CaH_2$. ¹H NMR and ¹³C NMR spectra were recorded with Me₄Si as internal reference. Mass spectrometry was carried out using chemical ionization (CI) methods. High-resolution masses were recorded with a double-focusing sector-field as analyzer (mass resolution up to 50.000) or an orbitrap as mass analyzer (HRMS of compounds 17, 19, and 23).

General Procedure for Reductive Aminations (General Procedure a). The amino compound (1.0 equiv) and the aldehyde (1.0 equiv) were dissolved in dry 1,2-DCE and then treated with solid sodium triacetoxyborohydride (1.5 equiv). The reaction mixture was stirred at room temperature, and completion of the reaction was easily monitored by TLC (using DCM/MeOH 9:1). Most of the reactions were already complete after 2 or 3 h. The reaction mixture was then quenched with saturated sodium hydrogen carbonate solution which was then extracted three times with DCM. The combined organic phases were dried over sodium sulfate, and the solvent was removed under reduced pressure with a rotatory evaporator. The products were purified either via distillation (parent secondary amines of compounds 9, 14, and 22) or via column chromatography (parent secondary amines of compounds 6−8, 10, 13, 15−21, 25) or used without further purification (parent amines of 11, 12, 23, and 24) in the following acylation reaction.

General Procedure for the Acylation with TBTU and the Carboxylic Acid (General Procedure b). The carboxylic acid (1.5 equiv, either benzoic acid or 3-benzoyl propionic acid) and TBTU (1.4 equiv) were suspended in DCM, and DIEA (6.0 equiv) was added. After 10 min, the secondary amine (1.0 equiv) was added. After stirring overnight at rt, the mixture was washed with water $2 \times$ and dried over $Na₂SO₄$. The organic solvent was evaporated, and the crude product was purified via flash chromatography (silica gel using $CH_2Cl_2/MeOH$ 98:2 \rightarrow 85:15 or aluminum oxide with CH/EE mixtures).

General Procedure for the Acylation Using Benzoyl Chloride and NEt₃ (General Procedure c). The secondary amine (1.0 equiv) and NEt₃ (3.0 equiv) were dissolved in DCM and cooled to 0 °C.

Scheme 2. Overview of Synthesized Chelating Amides

The Journal of Organic Chemistry **Article** 3 and 2008 and 20

Scheme 2. continued

Scheme 2. continued

Scheme 2. continued

 a Key: (a) NaBH(OAc)₃ (1.40 equiv), 1,2-DCE, rt; (b) TBTU (1.40 equiv), benzoic acid (1.40 equiv), DIEA (6.00 equiv), DCM, rt; (c) benzoyl chloride (3.00 equiv), NEt₃ (3.00 equiv), DCM, rt; (d) NH₄Cl (2.00 equiv), MeOH/water, reflux; (e) hydroxylammonium hydrochloride (1.00 equiv), rt; (f) Zn (5.00 equiv), HCl (2 M), reflux; (g) DMAP (0.09 equiv), DCC (1.00 equiv), MeOH (3.00 equiv), DCM, rt; (h) pyridinium-2 carboxaldehyde (1.30 equiv), CuI (0.10 equiv), NEt₃ (0.40 equiv), MeCN, rt; (i) MeLi (1.00 equiv), Et₂O, -78 °C to rt; KMnO₄, acetone, 2 h, rt; (j) SeO₂ (1.50 equiv), dioxane, reflux; (k) diethyl malonate (1.55 equiv), NaH (1.04 equiv), toluene, 120 °C; (l) H₂SO₄ (50%), 120 °C; (m) I₂ (1.00 equiv), TFA (3.00 equiv), t-BuI (0.40 equiv), DMSO, 170 °C; (n) H₂ (1 bar), Pd/C (10%), MeOH/toluene (1:2), rt; (o) n-BuLi (1.00 equiv), npentane, 1 h, −78 °C; 1 h, 0 °C; 1 h, rt.

Benzoyl chloride (3.0 equiv) was then added dropwise within 15 min. After removal of the ice bath, the reaction mixture was stirred at room temperature overnight. The reaction was then quenched with water, and the organic phase was washed with water one more time. After the mixture was dried over sodium sulfate, the solvent was removed under reduced pressure and the compounds were purified via column chromatography (over silica gel using CH₂Cl₂/MeOH 98:2 \rightarrow 85:15 or over basic aluminum oxide using CH/EE 4:1 \rightarrow EE \rightarrow EE/EtOH 98:2).

General Procedure of the Cleavage of the Chelating Amide to the Methyl Ester with $Cu(OTF)_{2}$ at rt. To a solution of the chelating amide (1.0 equiv) in MeOH (volume was added in order to get a concentration of 0.28 M) was added $Cu(OTf)_{2}$ (1.1 equiv), and the resulting mixture was stirred at rt. During the reaction, samples for HPLC measurements were taken with Hamilton syringes (10 μ L, diluted with 100 μ L of phenol standard solution and 180 μ L of MeOH). Alternatively, the reaction was performed at 50 °C and/or in the presence of DIEA (1.1 equiv). The samples for the HPLC measurements were taken in an analytical way.

Analytical Data. Benzylbis[(2-pyridyl)methyl]amine (5). Benzamide 5 was synthesized according to the general procedure b described above. The resulting dark oil was purified via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/MeOH 95:5). The resulting yellow gum (0.97 g, 3.2 mmol, 80%) crystallized overnight.

Benzylbis[(2-pyridyl)methyl]amine (5): mp 58–59 °C; ¹H NMR (CDCl₃, 300 MHz) δ = 4.70 (s, 2H), 4.90 (s, 2H), 7.15 – 7.20 (m, 3H), 7.32−7.45 (m, 4H), 7.53 −7.57 (m, 2H), 7.66 (m, 2 H), 8.55 (m, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ = 50.5, 54.6, 121.4, 122.4, 122.5, 127.0, 128.5, 129.8, 136.0, 136.8, 149.4, 149.9, 156.8, 157.2, 172.6 ppm, MS (CI, NH₃): $m/z = 304.1$ (100) [(M + 1)⁺]; HRMS (CI, \overline{NH}_3) m/z $[(M + 1)^+]$ calcd for $C_{19}H_{17}N_3O$ 303.13716, found 303.13720. Anal. Calcd for C₁₉H₁₇N₃O: C, 75.23; H, 5.65; N, 13.85. Found: C, 75.06; H, 5.62; N, 13.81.

Figure 1. (a) Methanolysis of benzamide 5 at rt, at 50 °C or/and in the presence of DIEA. (b) Methanolysis of benzamide 9 at rt, at 50 $^{\circ}$ C or/and in the presence of DIEA.

N-Propyl-N-pyridin-2-ylmethylbenzamide (6). The precursor amine 6a was synthesized by the general procedure a described above. A yellow liquid (0.93 g, 6.2 mmol, 72%) with the following analytical data was obtained after purification with column chromatography (silica gel, DCM/MeOH 98:2 → DCM/MeOH 80:20).

 $N\left(\left(\text{Pyridin-2-yl}\right) \text{methyl}\right)$ propan-1-amine $\left(6a\right) :\text{ }^{12}$ ¹H NMR $\left(\text{CDCl}_{3}\right)$ 300 MHz) δ = 0.95 (t, 3H, J = 7.4 Hz), 1.59–1.73 (m, 2H), 2.76 (t, 2H, J = 7.4 Hz), 4.07 (s, 2H), 7.22 (m_c, 1H), 7.3[3 \(](#page-11-0)d, 1H, J = 7.7 Hz), 7.69 (ddd, 1H, J = 7.7 Hz, J = 7.7 Hz, J = 1.8 Hz), 8.57 (m_c, 1H); ¹³C

100

80

60

40

20

% Cleavage

Scheme 3. Synthetic Steps for the Preparation of Aliphatic Amides 26 and 27

Figure 2. Methanolysis of 26 and 27 in the presence of $Cu(OTf)$ ₂ at rt or at 50 °C in the presence of $Cu(OTf)_{2}$ and DIEA.

NMR (CDCl₃, 100 MHz) $\delta = 11.8, 23.1, 51.5, 55.1, 121.9, 122.3$ 136.5, 149.3, 159.6; MS (CI, NH₃) $m/z = 151.1$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₉H₁₄N₂ 151.12352, found 151.12370.

Amine 6a was then acylated according to literature procedure b, and benzamide 6 was obtained as a yellow gum (0.74 g, 2.9 mmol, 48%) after purification via column chromatography (silica gel, DCM/MeOH 99:1 \rightarrow DCM/MeOH 98:2).

N-Propyl-N-pyridine-2-ylmethyl-benzamide $({\bf 6})$: $^1{\rm H}$ NMR $({\rm CDCl}_3,$ 300 MHz) δ = 0.63−1.00 (m, 3H), 1.48−1.77 (m, 2H), 3.19−3.54 (m, 2H), 4.58−4.96 (m, 2H), 7.20 (m_c, 1H), 7.29−7.47 (m, 6H), 7.68 (ddd, 1H, J = 7.7 Hz, J = 7.7 Hz, J = 1.5 Hz), 8.55 (m_c, 1H); ¹³C NMR $(CDCl_3, 100 MHz)$ δ = 11.2, 21.6, 47.1, 50.2, 51.2, 54.4, 121.1, 122.4, 126.5, 128.4, 129.4, 136.5, 136.9, 149.3, 157.4, 172.3; MS (CI, NH3)

Figure 3. Comparison of the cleavage rate of 5 and 9 with CuCl₂ and $Cu(OTf)$ ₂ at rt.

 $m/z = 255.1$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for $C_{16}H_{18}N_2O$ 255.14974, found 255.14930.

N-(2-Methoxyethyl)-N-pyridin-2-ylmethylbenzamide (7). The precursor amine 7a was synthesized according to the general procedure a) described above. A yellow liquid (0.79 g, 4.7 mmol, 50%) with the following analytical data was obtained after purification with column chromatography (silica gel, DCM/MeOH 98:2 → DCM/MeOH 90:10).

2-Methoxy-N-((pyridin-2-yl)methyl)ethanamine (7a):¹³ ¹H NMR (CDCl₃, 300 MHz) δ = 2.64 (br s, 1H), 2.87 (t, 2H, J = 5.2 Hz), 3.36 $(s, 3H)$, 3.55 (t, 2H, J = 5.2 Hz), 3.93 (s, 2H), 7.16 (m_o, [1H](#page-11-0)), 7.32 (d, 1H, $J = 7.7$ Hz), 7.63 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 8.55 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 49.0, 55.2, 59.0, 72.0, 122.1, 136.6, 159.4, 159.6; MS (CI, NH₃) $m/z = 167.1$ (100) $[(M + 1)^+]$.

Amine 7a was then acylated according to literature procedure c, and benzamide 7 was obtained as a brown gum (0.67 g, 2.5 mmol, 85%) after purification via column chromatography (silica gel, DCM/MeOH 98:2).

N-(2-Methoxyethyl)-N-pyridin-2-ylmethylbenzamide (7): ¹H NMR (CDCl₃, 300 MHz) δ = 3.14–3.27 (m, 2H), 3.30–3.43 (m, 2H), $3.47 - 3.79$ (m, 3H), $4.69 - 5.02$ (m, 2H), 7.19 (m_c, 1H), $7.28 - 7.75$ (m, 7H), 8.56 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.3, 49.1, 50.7, 55.7, 58.9, 70.0, 122.5, 127.0, 128.4, 129.6, 136.4, 172.6; MS (CI, NH₃) $m/z = 271.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + $[1)^+$] calcd for $C_{16}H_{18}N_2O_2$ 271.14520, found 271.14465.

N-(2-Methylsulfanylethyl)-N-pyridine-2-ylmethylbenzamide (8). The precursor amine 8a was synthesized according to the general procedure a described above. A dark yellow liquid (0.210 g, 1.14 mmol, 41%) with the following analytical data was obtained after purification with column chromatography (silica gel, DCM/MeOH $98:2 \rightarrow \text{DCM/MeOH} 90:10$).

2-(Methylthio)-N-((pyridin-2-yl)methyl)ethanamine $(8a):^{14}$ ¹H NMR (CDCl₃, 300 MHz) δ = 2.09 (s, 3H), 2.41 (br s, 1H), 2.71 (t, 2H, J = 6.5 Hz), 2.89 (t, 2H, J = 6.5 Hz), 3.96 (s, 2H), 7.[16](#page-11-0) (m_c, 1H), 7.33 (d, 1H, $J = 7.7$ Hz), 7.65 (dd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz), 8.56 (d, 1H, J = 4.3 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 15.4, 34.4, 48.7, 54.9, 122.2, 136.5, 149.5, 159.6; MS (CI, NH₃) $m/z = 183.1$ (100) $[(M + 1)⁺]$.

This amine was then acylated according to the general procedure c, and benzamide 8 was obtained as a brown gum (0.26 g, 0.92 mmol, 79%) after purification via column chromatography (silica gel, DCM/ MeOH 98:2 \rightarrow DCM/MeOH 95:5).

 N - $(2$ -Methylsulfanylethyl $)$ - N -pyridin-2-ylmethylbenzamide (8) : 1 H NMR (300 MHz, CDCl₃): δ = 1.96 (m_c, 3H), 2.70 (m_c, 2H), 3.63 (m_c, 2H), 4.79 (m_c, 2H), 7.12–7.24 (m, 1H), 7.27–7.66 (m, 6H), 7.66– 7.84 (m, 1H), 8.57 (m_c, 1H); ¹³C NMR (100 MHz, CDCl₃): δ = 15.5, 32.5, 45.0, 48.9, 50.3, 55.3, 122.9, 126.9, 128.6, 129.8, 134.6, 149.9, 136.2, 157.0, 172.4; MS (CI, NH₃) $m/z = 287.0$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₁₆H₁₈N₂OS 287.12180, found 287.12181.

N-(2-Dimethylaminoethyl)-N-pyridin-2-ylmethylbenzamide (9). The precursor amine 9a was synthesized by the general procedure a described above. A liquid with the following analytical data was obtained after distillation $(bp_{6.5*10}-p_{\text{max}} = 95-100 \text{ °C})$ as a yellow liquid (2.42 g, 13.5 mmol, 58%).

 N^1 , N^1 -Dimethyl- N^2 - $(($ pyridin-2-yl $)$ methyl $)$ ethane-1,2-diamine $(9a):^{15}$ ¹H NMR (CDCl₃, 300 MHz) δ = 2.28 (s, 6H), 2.58 (t, 2H, J = 6.2 Hz), 2.82 (t, 2H, $J = 6.2$ Hz), 3.96 (s, 2H), 7.16 (ddd, 1H, $J = 7.7$ Hz, J [=](#page-11-0) 4.8 Hz, J = 1.2 Hz), 7.32–7.34 (m, 1H), 7.64 (ddd, 1H, J = 7.7 Hz, J = 7.7 Hz, J = 1.8 Hz), 8.54 (m_o, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.7, 47.0, 55.5, 59.3, 122.0, 122.3, 136.5, 149.3, 159.9; MS (CI, NH₃) $m/z = 180.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{10}H_{17}N_3$ 180.15007, found 180.15020.

Amine 9a was then acylated according to the general procedure b, and benzamide 9 was obtained as a yellow gum (0.54 g, 1.9 mmol, 57%) after purification with column chromatography (silica gel, $DCM/MeOH$ 98:2 \rightarrow DCM/MeOH 80:20).

^N-(2-Dimethylamino-ethyl)-N-pyridine-2-ylmethylbenzamide (9): ¹ ¹H NMR (CDCl₃, 300 MHz) δ = 1.94–2.36 (m, 6H), 2.35–2.70 (m, 2H), 3.32−3.71 (m, 2H), 4.64−4.99 (m, 2H), 7.19 (m_o 1H), 7.29−7.43 (m, 4H), 7.44−7.47 (m, 2H), 7.67 (m, 1H), 8.55 (m_o, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.7, 122.6, 126.9, 128.5, 129.6, 137.0, 172.5; MS (CI, NH₃) $m/z = 284.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₁₇H₂₁N₃O 284.17629, found 284.17640.

N-(1-Pyridin-2-yl)ethyl)-N-((pyridin-2-yl)methyl)benzamide (10). The precursor amine 10a was synthesized according to the general procedure a described above. A brown liquid (2.08 g, 9.6 mmol, 96%) with the following analytical data was obtained without further purification.

1-(Pyridin-2-yl)-N-((pyridin-2-yl)methyl)ethanamine (10a):¹⁶¹H NMR (CDCl₃, 400 MHz) $\delta = 1.43$ (d, 3H, J = 6.7 Hz), 3.10 (br s, 1H), 3.77 (s, 2H, CH₂), 3.94 (q, 1H, J = 6.7 Hz), 7.08–7.14 ([m, 2](#page-11-0)H), 7.25 (d, 1H, J = 7.7 Hz), 7.36 (ddd, 1H, J = 7.8 Hz, J = 7.8 Hz, J = 1.1

Hz), 7.58 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.9$ Hz), 7.62 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.9$ Hz), 8.53 (m_c, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ = 22.9, 53.1, 59.2, 121.1, 121.9, 122.0, 122.3, 136.4, 136.6, 149.2, 149.3, 159.7, 164.4; MS (CI, NH₃) $m/z = 214.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₁₃H₁₅N₃ 214.13442, found 214.13420.

This amine was then acylated according to general procedure b, and 10 was obtained as a yellow gum (1.29 g, 4.05 mmol, 83%) after purification via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/MeOH 90:10). The yellow gum crystallized after several days.

 $N-(1-Pyridin-2-yl)$ ethyl $-N-(yyridin-2-yl)$ methyl ℓ benzamide (10): ¹H NMR (CDCl₃, 400 MHz) δ = 1.58 (d, 3H, J = 5.5 Hz), 4.32–6.10 (m, 3H), 6.90–7.70 (m, 11H), 8.47 (m_o, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ = 17.7, 47.8, 59.3, 121.6, 121.7, 122.4, 126.7, 128.7, 129.6, 136.5, 148.7, 149.5, 158.5, 172.9; MS (CI, NH₃) $m/z = 318.2$ (100) $[(M + 1)^+]$. Anal. Calcd for C₂₀H₁₉N₃O: C, 75.69; H, 6.03; N, 13.24. Found: C, 75.60; H, 6.14; N, 13.32.

N-(1-Methyl-1H-imidazol-4-ylmethyl)-N-pyridin-2-ylmethylbenzamide (11). The precursor amine 11a was synthesized according to the general procedure a described above. The pale yellow liquid 11a was then acylated according to the general procedure b, and 11 was obtained as a brown gum (0.27 g, 0.8 mmol, 83% over two steps) after purification via column chromatography (silica gel, DCM/MeOH 98:2 → DCM/MeOH 90:10). N-(1-Methyl-1H-imidazol-4-ylmethyl)-Npyridin-2-ylmethylbenzamide (11): ¹H NMR (CDCl₃, 300 MHz) δ = 3.64 (s, 3H), 4.36−4.67 (m, 2H), 4.67−4.95 (m, 2H), 6.70−6.99 (m, 1H), 7.14−7.25 (m, 1H), 7.28−7.51 (m, 6H), 7.64−7.72 (m, 2H), 8.51−8.59 (m, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 33.5, 42.6, 49.6, 118.5, 119.3, 121.45, 122.3, 127.0, 127.4, 128.4, 129.4, 136.8, 138.0, 138.3, 149.2, 149.8; MS (CI, NH₃) $m/z = 307.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₁₈H₁₈N₄O 307.15589, found 307.15570.

N-Pyridin-2-ylmethyl-N-quinolin-2-ylmethylbenzamide (12). The precursor amine 12a was synthesized according to the general procedure a described above. Amine 12a was then acylated according to literature procedure b, and 12 was obtained as a brown gum (0.44 g, 1.23 mmol, 46% over two steps) after purification via column chromatography (silica gel, DCM/MeOH 98:2 → DCM/MeOH 97:3). ^N-Pyridin-2-ylmethyl-N-quinolin-2-ylmethylbenzamide (12): ¹ ¹H NMR (CDCl₃, 300 MHz) δ = 4.69–4.90 (m, 2H), 4.94–5.11 $(m, 2H)$, 7.15 $(m_c, 1H)$, 7.17–7.28 $(m, 1H)$, 7.28–7.39 $(m, 3H)$, 7.45 $(m_c, 1H)$, 7.48−7.73 (m, 6H), 7.80 (dd, 1H, J = 8.1 Hz, J = 1.4 Hz), 7.99−8.05 (m, 1H), 8.09−8.16 (m, 1H), 8.52 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 48.9, 49.2, 52.7, 53.2, 117.2, 118.5, 119.6, 120.4, 120.7, 124.5, 125.1, 125.5, 125.7, 126.5, 127.3, 127.6, 127.8, 134.2, 134.8, 135.0, 147.4, 147.9, 170.9; MS (CI, NH₃) $m/z = 354.1$ (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{23}H_{19}N_3O$ 354.16064, found 354.16080.

N-Furan-2-yl-methyl-N-pyridin-2-ylmethylbenzamide (13). The precursor amine 13a was synthesized according to the general procedure a described above. A brown liquid (0.80 g, 4.25 mmol, 46%) with the following analytical data was obtained after purification with column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/ MeOH 85:15).

(Furan-2-yl)-N-((pyridin-2-yl)methyl)methanamine (13a): ¹H NMR (CDCl₃, 300 MHz) δ = 2.78 (br s, 1H), 3.85 (s, 2H), 3.93 $(s, 2H)$, 6.20 (dd, 1H, J = 3.1 Hz, J = 0.7 Hz), 6.30 (dd, 1H, J = 3.1 Hz, $J = 1.8$ Hz), 7.16 (m_c, 1H), 7.28 (d, 1H, $J = 7.7$ Hz), 7.37 (dd, 1H, $J =$ 1.8 Hz, $J = 0.7$ Hz), 7.63 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 8.56 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.8, 54.1, 107.3, 110.2, 122.3, 136.6, 142.0, 149.4, 153.6, 159.4; MS (CI, NH₃): $m/z =$ 189.2 (100) $[(M + 1)^+]$.

Amine 13a was then acylated according to literature procedure b, and 13 was obtained as a brown gum (0.59 g, 2.0 mmol, 95%) after purification via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/MeOH 90:10).

 N -Furan-2-yl-methyl- N -pyridin-2-ylmethyl-benzamide (13) : $^1\mathrm{H}$ NMR (CDCl₃, 300 MHz) δ = 4.58 (m_c, 2H), 4.80 (m_c, 2H), 6.15− 6.32 (m, 2H), 7.20 (dd, 1H, J = 7.4 Hz, J = 4.9 Hz), 7.31−7.76 (m, 8H), 8.57 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 46.4, 49.7, 109.2, 110.4, 122.4, 127.1, 128.3, 129.8, 132.9, 135.9, 142.7, 149.2; MS $(CI, NH₃) = 293.0 (100) [(M + 1)⁺]$; HRMS $(CI, NH₃) m/z [(M + 1)⁺]$ 1)⁺] calcd for $C_{18}H_{16}N_2O_2$ 293.12950, found 293.12900.

N-(Pyridin-2-yl)-N-((pyridin-2-yl)methyl)benzamide (14). The precursor amine 14a was synthesized by the general procedure a described above. A yellow liquid (1.53 g, 8.27 mmol, 75%) with the following analytical data was obtained after purification with column chromatography (silica gel, DCM/MeOH 98:2 → DCM/MeOH 95:5).

 N , N - $(2$ -Pyridyl $)(2$ -pyridylmethyl $)$ amine $(14a):^{17,18}$ ¹H NMR (CDCl₃, 400 MHz) δ = 4.63 (s, 2H), 4.76 (s, 1H), 6.46 (d, 1H, J = 8.5 Hz), 6.58 (ddd, 1H, J = 6.1 Hz, J = 5.2 Hz, J = [0.9 H](#page-11-0)z), 6.77 (s, 1H), 7.36 (d, 1H, $J = 7.7$ Hz), 7.42 (ddd, 1H, $J = 8.6$ Hz, $J = 7.4$ Hz, J $= 2.0$ Hz), 7.64 (ddd, 1H, J = 7.6 Hz, J = 7.6 Hz J = 1.8 Hz), 8.02 (m_c, 1H), 8.55 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 47.4, 108.2, 113.1, 121.8, 122.3, 136.8, 138.0, 148.6, 149.2, 157.9, 158.1 ppm.

This amine was then acylated according to the general procedure c and 14 was obtained as a crystalline powder (0.82 g, 2.84 mmol, 70%) after purification via column chromatography (aluminum oxide, CH/ EE 4:1).

N-(Pyridin-2-yl)-N-((pyridin-2-yl)methyl)benzamide (14): ¹H NMR (CDCl₃, 400 MHz) δ = 5.45 (s, 2H), 6.92 (ddd, 1H, J = 8.1 Hz, $J = 8.1$ Hz, $J = 0.9$ Hz), 6.99 (ddd, 1H, $J = 7.5$ Hz, $J = 4.9$ Hz, $J =$ 1.0 Hz), 7.13 (m_c, 1H), 7.19–7.24 (m, 2H), 7.28–7.33 (m, 1H), 7.39−7.44 (m, 3H), 7.46 (ddd, 1H, J = 7.9 Hz, J = 7.9 Hz, J = 1.0 Hz), 7.63 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 8.36 (ddd, 1H, $J =$ 4.9 Hz, $J = 1.9$ Hz, $J = 0.9$ Hz), 8.50 (ddd, 1H, $J = 4.1$ Hz, $J = 1.9$ Hz, J $= 1.0$ Hz); ¹³C NMR (CDCl₃, 100 MHz) $\delta = 53.6$, 121.0, 122.0, 122.1, 128.1, 128.8, 130.4, 136.0, 136.7, 137.3, 148.8, 149.2, 156.2, 157.7; MS (CI, NH₃) $m/z = 290.1$ (100) $[(M + 1)^+]$. Anal. Calcd for C18H15N3O: C, 74.72; H, 5.23; N, 14.52. Found: C, 74.50; H, 5.24; N, 14.50.

N-(2-(Pyridin-2-yl)ethyl)-N-((pyridin-2-yl)methyl)benzamide (15). Benzamide 15 was synthesized via a multistep reaction. The first step was the addition of ammonium chloride to 2-vinylpyridine which was commercially available.¹⁹ The primary amine 15a was obtained as a colorless liquid (1.76 g, 14.3 mmol, 29%) after distillation (b p_{16mbar} = 88 \degree C).

2-(Pyridin-2-yl)etha[nam](#page-11-0)ine $(15a):^{20}$ ¹H NMR (CDCl₃, 400 MHz) δ = 1.54 (s, 2H), 2.91 (t, 2H, J = 6.7 Hz), 3.10 (t, 2H, J = 6.7 Hz), 7.09 $(ddd, 1H, J = 7.5 Hz, J = 6.1 Hz, J = 1.1 Hz$ $(ddd, 1H, J = 7.5 Hz, J = 6.1 Hz, J = 1.1 Hz$ $(ddd, 1H, J = 7.5 Hz, J = 6.1 Hz, J = 1.1 Hz$), 7.14 $(ddd, 1H, J = 7.9$ Hz, $J = 0.9$ Hz, $J = 0.9$ Hz), 7.57 (ddd, 1H, $J = 7.6$ Hz, $J = 7.2$ Hz, $J =$ 1.9 Hz), 8.51 (ddd, 1H, J = 4.9 Hz, J = 1.9 Hz, J = 1.0 Hz); ¹³C NMR $(CDCl_3, 100 MHz)$ δ = 41.9, 121.2, 123.3, 136.3, 149.3, 160.0; MS (CI, NH₃) $m/z = 123.1$ (100) [(M + 1)⁺].

The secondary amine 15b was then synthesized via the general procedure a described above. 2-Pyridine-carboxaldehyde was used as the carbonyl compound. The secondary amine was obtained as a yellow liquid (1.47 g, 6.95 mmol, 97%) without further purification.

2-(Pyridin-2-yl)-N-((pyridin-2-yl)methyl)ethanamine $(15b)$:^{21,22} ¹H NMR (CDCl₃, 400 MHz) δ = 2.01 (s, 1H), 3.16 (m, 4H), 4.10 $(s, 2H)$, 7.13–7.22 (m, 3H), 7.37 (ddd, 1H, J = 7.7 Hz, J = 1.0 [Hz\),](#page-11-0) 7.62 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 7.67 (ddd, 1H, $J =$ 7.7 Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 8.51 (ddd, $J = 4.9$ Hz, $J = 1.8$ Hz, $J =$ 0.9 Hz), 8.54 (ddd, 1H, J = 4.9 Hz, J = 1.8 Hz, J = 0.9 Hz); ¹³C NMR $(CDCl₃, 100 MHz)$ δ = 35.8, 48.3, 55.7, 121.8, 122.8, 122.9, 123.5, 136.8, 136.9, 149.1, 149.5, 156.3, 159.4; MS (CI, NH₃) $m/z = 214.1$ (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{13}H_{15}N_3$ 214.13442, found 214.13430.

The secondary amine 15b was then acylated according to general procedure b. The benzamide 15 was obtained as yellow gum (0.55 g, 1.74 mmol, 62%) after purification via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/MeOH 97:3).

N-(2-(Pyridin-2-yl)ethyl)-N-((pyridin-2-yl)methyl)benzamide (15): ¹H NMR (CDCl₃, 400 MHz) δ = 3.12 (m_c, 2H), 3.84 (m_c, 2H), 4.70 (m_c, 2H), 6.84–7.71 (m, 11H), 8.50 (m_c, 2H); ¹³C NMR $(CDCl_3, 100 MHz)$ δ = 35.6, 37.0, 45.9, 49.4, 50.2, 55.0, 121.1, 121.7, 122.4, 124.0, 126.7, 128.2, 128.5, 129.5, 129.9, 136.4, 136.7, 149.0, 149.8; MS (CI, NH₃) $m/z = 318.2$ (100) [(M + 1)⁺]; HRMS (CI,

NH₃) m/z [(M + 1)⁺] calcd for C₂₀H₁₉N₃O 318.16064, found 318.16050.

N,N-Bis(2-(pyridin-2-yl)ethyl)benzamide (16). The parent secondary amine of benzamide 16 was prepared according to a literature process²³ in a two-step reaction. In the first step, hydroxylammonium hydrochloride was added to 2-vinylpyridine. The hydroxylamine derivat[ive](#page-11-0) 16a (2.17 g, 9.11 mmol, 31%) was obtained after recrystallization (n-hexane/DCM) as a brown solid.

Bis[2-(pyridin-2-ylethyl)]hydroxylamine (16a): mp 100 °C; ¹H NMR (CDCl₃, 400 MHz) δ = 3.10 (s, 8H), 7.06 (ddd, 2H, J = 6.2 Hz, $J = 5.0$ Hz, $J = 1.2$ Hz), 7.12 (ddd, $2H, J = 7.8$ Hz, $J = 1.0$ Hz, $J = 1.0$ Hz), 7.53 (ddd, 2H, J = 7.7 Hz, J = 7.7 Hz, J = 1.9 Hz), 8.44 (ddd, 2H, $J = 4.9$ Hz, $J = 1.9$ Hz, $J = 1.0$ Hz); ¹³C NMR (CDCl₃, 100 MHz) $\delta =$ 36.0, 60.0, 121.2, 123.5, 136.5, 148.9, 160.4 ppm.

In the second step, 16a was reduced to the corresponding amine with zinc and hydrochloric acid. After the solution was stirred at 95 °C for 90 min, the suspension was allowed to cool down and made basic with a solution of sodium hydroxide. The secondary amine 16b was obtained after extraction with DCM (3×) as yellow oil (0.70 g, 2.35 mmol, 97%): ¹H NMR (CDCl₃, 400 MHz) δ = 2.16 (br s, 1H), 3.00 $(t, 4H, J = 6.5 Hz)$, 3.08 $(t, 4H, J = 6.5 Hz)$, 7.10 $(dd, 2H, J = 5.8 Hz, J$ $= 5.8$ Hz), 7.15 (d, 2H, J = 7.7 Hz), 7.57 (ddd, 2H, J = 7.8 Hz, J = 7.8 Hz, $J = 1.8$ Hz), 8.49 (d, 2H, $J = 4.8$ Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 49.2, 121.4, 123.4, 136.5, 149.4, 160.2 ppm.

Benzamide 16 was then synthesized according to general procedure b. Compound 16 was obtained as a yellow oil (0.32 g, 0.97 mmol, 57%) after purification via column chromatography (silica gel, DCM/ MeOH 99:1 \rightarrow 97:3).

N,N-Bis(2-(pyridin-2-yl)ethyl)benzamide (16): ¹H NMR (CDCl₃, 400 MHz) δ = 3.41 (m_c, 8H), 6.90 (br s, 1H), 7.04–7.20 (m, 4H), 7.26−7.36 (m, 4H), 7.58 (m_c, 2H), 8.47 (m_c, 2H); MS (CI, NH₃) m/z = 332.2 (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{21}H_{21}N_3O$ 332.17629, found 332.17630.

4-[5-[(Benzoylpyridin-2-ylmethylamino)methyl][1,2,3]triazol-1 yl]benzoic Acid Methyl Ester (17). Benzamide 17 was synthesized in a multistep reaction. Its triazol unit was built up via a $\begin{bmatrix} 2 & + & 3 \end{bmatrix}$ cycloaddition. The propargyl amine 17a was synthesized by a reductive amination step according to the general procedure a. The propargyl amine was a dark brown oil (0.91 g, 6.23 mmol, 33%) purified via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow 90:10). Prop-2-ynylpyridin-2-ylmethylamine $(17a):^{24}$ ¹H NMR $(CDCl₃, 300$ MHz) δ = 2.12 (br s, 1H), 2.24 (t, 1H, J = 2.4 Hz), 3.51 (d, 2H, J = 2.4 Hz), 4.01 (s, 2H), 7.17 (m_c, [1H](#page-11-0)), 7.33 (d, 1H, J = 7.7 Hz), 7.64 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 8.56 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 37.8, 53.7, 71.7, 81.9, 122.4, 136.5, 149.5, 159.2; MS (CI, NH₃) $m/z = 147.1$ (100) [(M + 1)⁺].

The azido building block 17b was prepared via a methylester formation using the commercially available carboxylic acid (1.0 equiv), dry MeOH, DMAP (0.09 equiv), and DCC (1.0 equiv). The reaction mixture was stirred at rt for 4 h. The methyl ester was isolated as a yellow solid (2.02 g, 11.4 mmol, 93%) after purification via column chromatography (silica gel, DCM/MeOH 99:1).

4-Azido-benzoic acid methyl ester $(17b)$: ¹H NMR $(CDCl_3$, 300 MHz) δ = 3.91 (s, 3H), 7.06 (m_c, 2H), 8.03 (m_c, 2H); ¹³C NMR $(CDCl_3, 100 MHz)$ δ = 35.0, 52.2, 118.9, 126.9, 131.5, 144.8, 166.4 ppm.

The $\lceil 2 + 3 \rceil$ -cycloaddition resulting in amine 17c was achieved by stirring azide 17b (1.3 equiv) and propargyl amine 17a (1.0 equiv) in CH₃CN in the presence of CuI (0.1 equiv) and NEt₃ (0.4 equiv) for 48 h. Amine 17c was obtained as a brown gum (0.15 g, 0.45 mmol, 51%) after purification via column chromatography (silica gel, DCM/ MeOH 99:1 \rightarrow DCM/MeOH 90:10).

4-[5-[[(Pyridin-2-ylmethyl)amino]methyl][1,2,3]triazol-1-yl] benzoic acid methyl ester (17c): ¹H NMR (CDCl₃, 300 MHz) δ = 2.53 (s, 1H), 3.96 (s, 3H), 4.04 (s, 2H), 4.11 (s, 2H), 7.18 (m_c, 1H), 7.34 (d, 1H, $J = 7.7$ Hz), 7.66 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, $J = 1.8$ Hz), 7.84 (m_c, 2H), 8.12 (s, 1H), 8.19 (m_c, 2H), 8.57 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 44.3, 52.5, 54.5, 119.9, 120.0, 122.2, 122.5, 130.2, 130.7, 131.4, 136.6, 140.3, 148.0, 149.5, 159.0, 166.0; MS (CI, NH₃) $m/z = 324.0$ (100) [(M + 1)⁺].

Amine 17c was then acylated according to the general procedure c, and benzamide 17 was isolated as a brown solid (0.12 g, 0.28 mmol, 77%) after purification via column chromatography (silica gel, DCM/ MeOH 98:2 \rightarrow DCM/MeOH 95:5).

4-[5-[(Benzoyl-pyridin-2-ylmethylamino)methyl][1,2,3]triazol-1 yl]benzoic acid methyl ester (17): mp 107 °C; ¹H NMR (CDCl₃, 300 MHz) δ = 3.97 (s, 3H), 4.75 (s, 2H), 4.85 (s, 2H), 7.24 (m_c, 1H), 7.31−7.89 (m, 7H), 7.87 (m_o, 2H), 8.21 (m_o, 2H), 8.29 (s, 1H), 8.63 $(m_c, 1H);$ ¹³C NMR (CDCl₃, 100 MHz) δ = 41.0, 52.5, 54.5, 119.9, 121.9, 122.0, 127.3, 128.5, 130.1, 131.4, 135.6, 136.9, 140.2, 145.2, 150.0, 156.4, 166.0, 172.7; MS (CI, NH₃) $m/z = 428.2$ (100) $[(M +$ 1)⁺]; HRMS (APCI, MeOH) m/z [(M + 1)⁺] calcd for $C_{24}H_{22}N_5O_3$ 428.17226, found 428.17220.

N-((Pyridin-2-yl)methyl)-N-(6-(pyridine-2-yl)pyridin-2-yl)methyl) benzamide (18). Benzamide 18 was synthesized in a multistep reaction using commercially available 2,2′-bipyridine as starting material. 2,2′-Bipyridine was methylated according to a literature process, 25 and 18a was obtained as a yellow gum (1.83 g, 10.6 mmol, 30%) after purification via column chromatography (silica gel, DCM/ MeOH [97](#page-11-0):3).

2-Methyl-6-(pyridin-2-yl)pyridine $(18a)$: ¹H NMR (CDCl₃, 400) MHz) δ = 2.61 (s, 3H), 7.13 (d, 1H, J = 7.7 Hz), 7.23–7.29 (m, 1H), 7.66 (t, 1H, J = 7.8 Hz), 7.74−7.81 (m, 1H), 8.15 (d, 1H, J = 7.6 Hz), 8.38 (m, 1H), 8.64–8.67 (m, 1H); ¹³C NMR (CDCl₃, 100 MHz) $\delta =$ 24.6, 118.1, 121.2, 123.3, 123.5, 136.8, 137.1, 149.2, 155.6, 156.5, 157.9 ppm.

Compound 18a was then oxidized to the corresponding carboxaldehyde 18b according to literature.²⁶ Carboxaldehyde 18b was then isolated as yellow powder (1.06 g, 5.76 mmol, 56%) after purification via column chromatography (sili[ca](#page-11-0) gel, DCM/EE 90:10). 6-(Pyridin-2-yl)pyridin-2-carbaldehyde $(18b)$: ¹H NMR $(CDCl_3$, 400 MHz) δ = 7.36 (ddd, 1H, J = 7.7 Hz, J = 4.8 Hz, J = 1.1 Hz), 7.87 (ddd, 1H, J = 7.8 Hz, J = 7.8 Hz, J = 1.8 Hz), 7.95−8.02 (m, 2H), 8.55 $(ddd, 1H, J = 7.8 Hz, J = 7.8 Hz, J = 1.0 Hz$), 8.66 $(dd, 1H, J = 6.7 Hz$, $\vec{J} = 2.3 \text{ Hz}$), 8.71 (ddd, 1H, $\vec{J} = 4.7 \text{ Hz}$, $\vec{J} = 1.8 \text{ Hz}$, $\vec{J} = 0.9 \text{ Hz}$), 10.17 (d, 1H, J = 0.7 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 121.4, 121.5, 124.4, 125.3, 137.2, 138.0, 149.3, 152.4, 155.0, 156.7, 193.7; MS (CI, NH₃): $m/z = 184.0$ (100) [(M + 1)⁺].

The secondary amine was then prepared via a reductive amination step according to general procedure a, and amine 18c was obtained as a brown oil (1.04 g, 3.75 mmol, 68%) without further purification.

(Pyridin-2-yl)-N-((6-(pyridin-2-yl)pyridin-2-yl)methyl) methanamine (18c): ¹H NMR (CDCl₃, 400 MHz) δ = 3.33 (br s, 1H), 4.07 (s, 2H), 4.09 (s, 2H), 7.16 (dd, 1H, J = 7.7 Hz, J = 4.9 Hz), 7.28 (ddd, 1H, $J = 7.5$ Hz, $J = 4.8$ Hz, $J = 1.2$ Hz), 7.35 (d, 1H, $J = 7.8$ Hz), 7.38 (d, 1H, $J = 7.8$ Hz), 7.64 (ddd, 1H, $J = 7.7$ Hz, $J = 7.7$ Hz, J = 1.8 Hz), 7.73−7.82 (m, 2H), 8.26 (d, 1H, J = 7.9 Hz), 8.45 (ddd, 1H, $J = 8.0$ Hz, $J = 8.0$ Hz, $J = 1.0$ Hz), 8.57 (ddd, 1H, $J = 4.8$ Hz, $J = 1.5$ Hz, $J = 0.8$ Hz), 8.66 (ddd, 1H, $J = 4.8$ Hz, $J = 1.7$ Hz, $J = 0.9$ Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 54.5, 54.6, 119.5, 121.3, 122.2, 122.3, 122.4, 123.7, 136.6, 136.9, 137.5, 149.2, 149.4, 155.7, 156.2, 158.4, 159.2; MS (CI, NH₃): $m/z = 277.2$ (100) [(M + 1)⁺].

Amine 18c was then acylated according to the general procedure c. Benzamide 18 was obtained as a yellow gum (0.98 g, 2.57 mmol, 66%) after purification via column chromatography (aluminum oxide, CH/ EE 4:1 \rightarrow CH/EE 1:1).

N-((Pyridin-2-yl)methyl)-N-(6-(pyridine-2-yl)pyridin-2-yl)methyl) benzamide (18): ¹H NMR (CDCl₃, 400 MHz) δ = 4.79 (s, 2H), 4.97 $(d, 2H, J = 10.8 Hz)$, 7.10−7.53 (m, 10H), 7.66 (ddd, 1H, J = 7.7 Hz, J $= 7.7$ Hz, $J = 1.9$ Hz), $7.73 - 7.86$ (m, 2H), 8.31 (d, 1H, $J = 7.8$ Hz), 8.36−8.46 (dd, 1H, J = 18.8 Hz, J = 8.0 Hz), 8.49−8.59 (m, 1H), 8.67 (ddd, 1H, $J = 4.8$ Hz, $J = 1.8$ Hz, $J = 0.9$ Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 50.5, 50.8, 54.6, 77.3, 119.6, 119.8, 121.3, 121.7, 122.5, 122.6, 123.0, 123.7, 124.0, 149.2, 127.0, 127.1, 128.4, 129.8, 136.2, 137.1, 137.7, 149.1, 149.9; MS (CI, NH₃) $m/z = 381.2$ (100) $[(M +$ 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₂₄H₁₈N₄O 381.17154, found 381.17130.

N-(5-Nitropyridin-2-ylmethyl)-N-pyridin-2-ylmethylbenzamide (19) and N-(5-Aminopyridin-2-ylmethyl)-N-pyridin-2-ylmethylbenzamide (20). Benzamide 19 was prepared in a multistep reaction sequence. In the first step, malonic acid diethylester was added dropwise to a suspension of sodium hydride (1.1 equiv). The resulting white suspension was then stirred for 30 min at 120 °C. Then, 2 chloro-5-nitropyridine (1.0 equiv) in toluene was added dropwise, and the reaction mixture stirred for 8 h at 120 °C. After the mixture was cooled to rt, the solvent was removed at reduced pressure using the rotatory evaporator. The oily residue was then resolved in sulfuric acid (50%), and the reaction mixture was refluxed for 8 h. After the mixture was cooled to rt and water was added, a concentrated sodium hydroxide solution was added slowly at 0 °C. Furthermore, a solution of saturated $NAHCO₃$ was added. After the water phases were extracted with $Et₂O$, the organic layers were dried over sodium sulfate and the picoline derivative 19a was obtained as a pink solid (2.15 g, 15.6 mmol, 39%) after purification via column chromatography (silica gel, DCM).

2-Methyl-5-nitropyridine $(19a).^{27}$ mp 102 °C; ¹H NMR (CDCl₃, 300 MHz) δ = 2.70 (s, 3H), 7.34 (d, 1H, J = 8.6 Hz), 8.36 (dd, 1H, J = 2.7 Hz, $J = 8.6$ $J = 8.6$ $J = 8.6$ Hz), 9.33 (d, 1H, $J = 2.7$ Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 24.9, 123.4, 131.3, 144.8, 165.5; MS (CI, NH₃) m/z = 139.1 (100) $[(M + 1)^+]$. Anal. Calcd for $C_6H_6N_2O_2$: C, 52.17; H, 4.38; N, 20.28. Found: C, 52.34; H, 4.47; N, 20.00.

The picoline derivative 19a was dissolved in DMSO, and iodine (1.0 equiv), tert-butyliodine (0.4 equiv), and TFA (3.0 equiv) were added. The resulting reaction mixture was then stirred for 3 h at 170 $^{\circ}$ C. After cooling down to rt, a NaHCO₃ solution (10%) and a $Na₂S₂O₃$ solution (1 M) were added. The combined water phases were extracted with DCM several times, and the combined organic phases were dried over sodium sulfate. The carbaldehyde 19b was obtained as yellow crystals (1.11 g, 7.23 mmol, 56%) after purification via column chromatography (silica gel, DCM).

5-Nitropyridine-2-carbaldehyde (19b):²⁸ mp 55 °C, ¹H NMR (CDCl₃, 300 MHz) δ = 8.16 (dd, 1H, I = 8.5 Hz, I = 0.6 Hz), 8.66 (d[d, 1](#page-11-0)H, $J = 8.5$ Hz, $J = 2.4$ Hz), 9.59 (d, 1H, $J = 2.4$ Hz), 10.18 (s, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 121.8, 132.6, 145.8, 146.1, 155.8, 191.3; MS (CI, NH₃) $m/z = 153.1$ (100) [(M + 1)⁺]. Anal. Calcd for $C_6H_4N_2O_3$: C, 47.38; H, 2.65; N, 18.42. Found: C, 46.91; H, 2.67; N, 18.76.

The secondary amine 19c was prepared according to the general procedure a, and 19c was obtained as a dark brown gum (0.15 g, 0.61 mmol, 52%) after purification via column chromatography (silica gel, $DCM/MeOH$ 98:2 \rightarrow $DCM/MeOH$ 90:10).

(5-Nitropyridine-2-ylmethyl)pyridin-2-ylmethylamine (19c): ¹ H NMR (CDCl₃, 300 MHz) δ = 4.02 (s, 2H), 4.14 (s, 2H), 7.19 (m_c, 1H), 7.32 (d, 1H, J = 7.8 Hz), 7.62−7.70 (m, 2H), 8.44 (dd, 1H, J = 8.6 Hz, J = 2.5 Hz), 8.58 (m_c, 1H), 9.38 (d, 1H, J = 2.5 Hz); ¹³C NMR $(CDCl₃, 100 MHz)$ δ = 54.6, 60.5, 122.3, 122.4, 131.6, 136.7, 143.2, 144.8, 149.5, 159.1, 166.9 ppm.

Benzamide 19 was prepared according to the general procedure c. Benzamide 19 was obtained as a brown gum (0.18 g, 0.52 mmol, 92%) after purification via column chromatography (silica gel, DCM/MeOH 98:2).

N-(5-Nitropyridin-2-ylmethyl)-N-pyridin-2-ylmethylbenzamide (19): ¹H NMR (CDCl₃, 300 MHz) δ = 4.74–4.86 (m, 2H), 4.87– 4.97 (m, 2H), 7.20 (m_c, 2H), 7.28–7.73 (m, 8H), 8.44 (m_c, 1H), 9.35 $(m_{\sigma} 1H);$ ¹³C NMR (CDCl₃, 100 MHz) δ = 50.9, 55.4, 121.8, 122.9, 127.2, 128.7, 130.2, 131.7, 135.5, 136.9, 143.4, 144.8, 150.0, 156.3, 164.0; MS (CI, NH₃) $m/z = 349.2$ (100) [(M + 1)⁺]; HRMS (APCI, MeOH) m/z $[(M + 1)^+]$ calcd for $C_{19}H_{16}N_4O_3$ 349.13007, found 349.13000. Anal. Calcd for C₁₉H₁₆N₄O₃: C, 65.51; H, 4.63; N, 16.08. Found: C, 65.12; H, 4.71; N, 16.18.

Benzamide 20 was prepared via a hydrogenation of benzamide 19. Benzamide 20 was dissolved in MeOH/toluene (1:2) and Pd/C was added. The resulting black suspension was then stirred for 24 h in a hydrogen atmosphere (1 bar). The suspension was then filtrated over kieselguhr and washed with MeOH. After the solvent was removed under reduced pressure and purification via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/MeOH 95:5), 20 was obtained as yellow gum (0.25 g, 0.78 mmol, 63%).

N-(5-Aminopyridin-2-ylmethyl)-N-pyridin-2-ylmethylbenzamide (20): ¹H NMR (CDCl₃, 300 MHz) δ = 3.56 (br s, 2H), 4.53 (m_c,

2H), 4.74 (m_c, 2H), 6.82–6.94 (m, 2H), 7.07–7.62 (m, 8 H) 7.94 $(m_c, 1H), 8.47 (m_c, 1H);$ ¹³C NMR (CDCl₃, 100 MHz) δ = 50.1, 59.1, 66.3, 122.3, 127.0, 128.4, 129.7, 136.2, 136.8, 137.2, 141.8, 145.7, 146.5, 149.3, 149.8, 156.8, 157.3; MS (CI, NH₃) $m/z = 319.2$ (100) $[(M + 1)^+]$, HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{19}H_{18}N_4O$ 319.15589, found 319.15660.

Methyl 6-((N-((Pyridine-2-yl)methyl)benzamido)methyl)pyridine-3-carboxylate (21). Benzamide 21 was prepared via a multistep reaction beginning with the oxidation of 6-methyl-nicotinic acid leading to carbaldehyde 21a. Carbaldehyde 21a was obtained as a pale yellow solid (1.54 g, 9.28 mmol, 62%) after purification via column chromatography (silica gel, CH/EE 3:1).

Methyl 6-formylpyridin-3-carboxylate $(21a).^{29}$ mp 107 °C; ¹H NMR (CDCl₃, 400 MHz) δ = 4.00 (s, 3H), 8.03 (dd, 1H, J = 8.1 Hz, J $= 0.9$ Hz), 8.46 (ddd, 1H, [J](#page-11-0) = 8.1 Hz, J = 2.0 Hz, J = 0.9 Hz), 9.36 (dd, 1H, $J = 2.0$ Hz, $J = 0.9$ Hz), 10.14 (d, 1H, $J = 0.9$ Hz); ¹³C NMR $(CDCl₃, 100 MHz)$ δ = 52.9, 121.2, 129.3, 138.4, 151.3, 155.0, 164.9, 192.7; MS (CI, NH₃) $m/z = 166.0$ (100) [(M + 1)⁺].

A reductive amination step according to the general procedure a using 21a as carbonyl compound and 2-picolylamine as amine lead to the secondary amine 21b which was obtained as dark brown oil (0.41 g, 1.59 mmol, 93%) without further purification.

Methyl 6-(((pyridine-2-yl)methylamino)methyl)pyridine-3-carboxylate (21b): ¹H NMR (CDCl₃, 400 MHz) δ = 3.94 (s, 3H), 4.01 (s, 2H), 4.07 (s, 2H), 7.17 (ddd, 1H, $J = 7.5$ Hz, $J = 4.9$ Hz, $J = 1.2$ Hz), 7.33 (d, 1H, $J = 7.9$ Hz), 7.47 (dd, 1H, $J = 8.2$ Hz, $J = 0.8$ Hz), 7.64 $(ddd, 1H, J = 7.7 Hz, J = 7.7 Hz, J = 1.9 Hz, 8.24 (dd, 1H, J = 8.1 Hz,$ $J = 2.2$ Hz), 8.56 (ddd, 1H, $J = 4.8$ Hz, $J = 1.7$ Hz, $J = 0.8$ Hz), 9.15 (dd, 1H, J = 2.2 Hz, J = 0.8 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 52.4, 54.5, 54.6, 121.8, 122.3, 122.5, 124.6, 136.6, 137.7, 149.5, 150.7, 159.0, 164.0, 165.9 ppm.

The secondary amine 21b was the acylated according to the general procedure c, and 21 was obtained as a pale yellow gum (0.33 g, 0.91 mmol, 58%) after purification via column chromatography (silica gel, DCM/MeOH 97:3).

Methyl 6-((N-((pyridine-2-yl)methyl)benzamido)methyl)pyridine-3-carboxylate (21): ¹H NMR (CDCl₃, 400 MHz) δ = 3.95 (s, 3H), 4.71 (br s, 1H), 4.80 (br s, 1H), 4.92 (br s, 2H), 7.13−7.59 (m, 8H), 7.70 (ddd, 1H, J = 6.5 Hz, J = 6.5 Hz, J = 3.3 Hz), 8.25 (m_c, 1H), 8.54 $(m_c, 1H)$, 8.54 (d, 1H, J = 12.4 Hz), 9.13 (s, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 48.7, 50.5, 52.8, 53.0, 119.2, 119.7, 120.1, 120.7, 125.0, 125.2, 126.6, 128.0, 133.8, 134.9, 135.9, 148.0, 148.7, 149.1, 154.6, 159.3, 159.8; MS (CI, NH₃) $m/z = 362.1$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for C₂₃H₂₁N₃O 362.15047, found 362.15040.

N,N-Bis(2-dimethylaminoethyl)benzamide (22). Compound 22a was prepared according to a literature procedure³⁰ followed by distillation (b p_{16mbar} = 90 °C) to obtain a colorless liquid (1.72 g, 10.8 mmol, 83%).

 N, N, N', N' -Tetramethyldiethylamine (22a): ¹H N[MR](#page-11-0) (CDCl₃, 300 MHz) δ = 2.22 (s, 12H), 2.41 (t, 4H, J = 6.4 Hz), 2.71 (t, 4H, J = 6.4 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.7, 47.8, 59.4; MS (CI, NH₃) $m/z = 160.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + $[1)^+$] calcd for $C_8H_{21}N_3$ 160.18150, found 160.18137.

Benzamide 22 was prepared out of amine 22a according to the general procedure b. Benzamide 22 was obtained as a yellow gum (0.57 g, 2.2 mmol, 65%) after purification via bulb to bulb distillation $(bp_{6.8*10}^{-2}mbar = 240 °C).$

 $N_\nu N$ -Bis(2-dimethylaminoethyl)benzamide (22): $^1{\rm H}$ NMR (CDCl₃, 300 MHz) δ = 1.98–2.35 (m, 12H), 2.35–2.64 (m, 4H), 3.29–3.69 (m, 4H), 7.39 (m, 5H); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.7, 47.0, 55.5, 59.3, 122.0, 122.3, 136.5, 149.3, 159.9; MS (CI, NH₃) $m/z =$ 264.2 (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{15}H_{25}N_3O$ 264.20810, found 264.20759.

N-(2-Dimethylamino-ethyl)-N-(1-methyl-1H-imidazol-4 ylmethyl)benzamide (23). The secondary amine 23a was prepared according to the general procedure a and than the secondary amine 23a was acylated according to the general procedure b. The benzamide 23 was obtained as a yellow gum (0.07 g, 0.2 mmol, 30% over two steps) after purification via column chromatography (aluminum oxide, $EE \rightarrow EE/EtOH$ 98:2).

N-(2-Dimethylaminoethyl)-N-(1-methyl-1H-imidazol-4-ylmethyl) benzamide (23): ¹H NMR (CDCl₃, 300 MHz) δ = 1.90–2.29 (m, 6H), 2.29−2.56 (m, 2H), 3.26−3.57 (m, 2H), 3.58 (s, 3H), 4.31−4.62 (m, 2H), 6.59–6.94 (m, 1H), 7.2–7.38 (m, 5H), 7.38–7.51 (m, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 42.6, 45.6, 46.9, 47.7, 56.6, 57.6, 117.9, 119.0, 126.6, 126.9, 128.3, 129.3, 136.8, 137.8, 138.8, 171.8; MS (pos ESI) $m/z = 287.3$ (100) $[(M + 1)^+]$; HRMS (APCI, MeOH/ NH_4) m/z $[(M + 1)^+]$ calcd for $C_{16}H_{22}N_4O$ 287.18730, found 287.18719.

N-(2-Dimethylaminoethyl)-N-quinolin-2-ylmethylbenzamide (24). Benzamide 24 was prepared via a reductive amination step according to general procedure a followed by an acylation according to general procedure b. Benzamide 24 was then obtained as a brown gum (0.16 g, 0.48 mmol, 20% over two steps) after purification via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow DCM/MeOH 85:15).

^N-(2-Dimethylaminoethyl)-N-quinolin-2-ylmethylbenzamide (24): ¹ ¹H NMR (CDCl₃, 300 MHz) δ = 2.71–2.86 (m, 2H), 3.41 (s, 6H), 3.97−4.15 (m, 1H), 4.88−5.06 (m, 2H) 7.12−7.31 (m, 3H), 7.23− 7.57 (m, 4H), 7.68−7.74 (m, 1H), 7.76−7.82 (m, 1H), 8.01−8.10 (m, 2H); ¹³C NMR (CDCl₃, 100 MHz) δ = 29.1, 40.9, 43.5, 53.6, 55.1, 62.7, 120.1, 126.6, 126.7, 126.8, 127.3, 127.6, 127.7, 128.3, 128.9, 129.8, 130.0, 135.3, 137.1, 147.6, 156.0, 173.4; MS (CI, NH₃) $m/z =$ 334.2 (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{21}H_{23}N_3O$ 334.19194, found 334.19250.

N-(2-Dimethylaminophenyl)-N-pyridin-2-ylmethylbenzamide (25). The precursor amine was synthesized by the general procedure a described above. A yellow liquid (0.58 g, 2.6 mmol, 71%) with the following analytical data was obtained after column chromatography (silica gel, CH/EE 5:1 \rightarrow CH/EE 2:1).

N,N-Dimethyl-N-pyridin-2-ylmethylbenzene-1,2-diamine (25a): ¹H NMR (CDCl₃, 300 MHz) δ = 2.70 (s, 6H), 4.50 (s, 2H), 6.52 (dd, 1H, $J = 8.0$ Hz, $J = 1.4$ Hz), 6.68 (ddd, 1H, $J = 7.6$ Hz, $J = 7.6$ Hz, $J = 1.4$ Hz), 6.93 (dddd, 1H, J = 7.4 Hz, J = 7.4 Hz, J = 1.5 Hz, J = 0.5 Hz), 7.05 (dd, 1H, $J = 7.7$ Hz, $J = 1.5$ Hz), 7.14 (m, 1H), 7.34 (m, 1H), 7.61 (ddd, 1H, J = 7.6 Hz, J = 7.6 Hz, J = 1.8 Hz), 8.58 (m_c, 1H); ¹³C NMR $(CDCl_3, 100 MHz)$ δ = 44.2, 50.0, 110.6, 117.0, 119.2, 121.2, 122.1, 124.8, 136.8, 140.7, 142.9, 149.4, 159.8; MS (CI, NH₃) $m/z = 228.2$ (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{14}H_{17}N_3$ 228.15007, found 228.14970.

The secondary amine 25a was then acylated according to the general procedure b, and 25 was isolated as a yellow gum (0.24 g, 0.7 mmol, 65%) after purification via column chromatography (silica gel, DCM/MeOH 98:2 \rightarrow 97:3).

^N-(2-Dimethylaminophenyl)-N-pyridin-2-ylmethylbenzamide (25): ¹ ¹H NMR (CDCl₃, 300 MHz) δ = 2.52 (s, 6H), 4.26 (d, 1H, J = 15.2 Hz), 5.89 (d, 1H, $J = 15.2$ Hz), 6.78 (dd, 1H, $J = 8.2$ Hz, $J = 1.4$ Hz), 6.84 (dddd, 1H, $J = 7.4$ Hz, $J = 7.4$ Hz, $J = 1.5$ Hz, $J = 0.3$ Hz), 7.07 (dddd, 1H, J = 7.4 Hz, J = 7.4 Hz, J = 1.6 Hz, J = 0.7 Hz), 7.10–7.23 (m, 5H), 7.38−7.42 (m, 2H), 7.50−7.54 (m, 1H), 7.67 (ddd, 1H, J = 7.7 Hz, J = 7.7 Hz, J = 1.8 Hz), 8.58 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 42.8, 54.7, 120.1, 122.1, 122.4, 122.5, 127.1, 127.9, 128.3, 182.8, 129.5, 136.2, 136.8, 148.5, 149.2, 158.1, 171.2; MS (CI, NH3) $m/z = 332.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + 1)⁺] calcd for $C_{21}H_{21}N_3O$ 332.17629, found 332.17640.

4-Oxo-4-phenyl-N,N-bis-pyridin-2-ylmethylbutyramide (26). Amine 2 was acylated according to the general procedure b, and the aliphatic amide 26 was obtained as a brown gum (0.52 g, 1.44 mmol, 82%) after purification with column chromatography (silica gel, $DCM/MeOH$ 98:2 \rightarrow $DCM/MeOH$ 90:10).

4-Oxo-4-phenyl-N,N-bis-pyridin-2-ylmethyl-butyramide $(26):$ ^{1}H NMR (CDCl₃, 300 MHz) δ = 2.93 (t, 2H, J = 6.4 Hz), 3.41 (t, 2H, J = 6.4 Hz), 4.78 (s, 2H), 4.82 (s, 2H), 7.14 (ddd, 1H, J = 7.6 Hz, J = 4.9 Hz, J = 1.2 Hz), 7.18 (ddd, 1H, J = 7.6 Hz, J = 4.9 Hz, J = 1.2 Hz), 7.31 (m_c, 2H), 7.41−7.47 (m, 2H), 7.51−7.56 (m, 2H), 7.62 (td, 1H, J $= 7.7$ Hz, $J = 1.8$ Hz), 7.69 (td, 1H, $J = 7.7$ Hz, $J = 1.8$ Hz), 7.98–8.02 $(m, 2H)$, 8.48 $(m_o 1H)$, 8.56 $(m_o 1H)$; ¹³C NMR (CDCl₃, 100 MHz) δ = 27.6, 34.1, 51.9, 53.2, 120.9, 122.3, 122.4, 128.1, 128.7, 133.0, 136.9, 149.2, 149.9, 156.8, 157.4, 172.8, 199.2; MS (CI, NH₃) $m/z =$ 360.2 (100) $[(M + 1)^+]$; HRMS (CI, NH₃) m/z $[(M + 1)^+]$ calcd for $C_{22}H_{21}N_3O_2$ 360.17120, found 360.17100.

N-(2-Dimethylaminoethyl)-4-oxo-4-phenyl-N-pyridin-2-ylmethylbutyramide (27). Amine 9a was acylated according to the general procedure b, and the aliphatic amide 27 was obtained as a brown gum (0.42 g, 1.22 mmol, 56%) after purification with column chromatography (silica gel, DCM/MeOH 98:2 → DCM/MeOH 80:20).

N-(2-Dimethylaminoethyl)-4-oxo-4-phenyl-N-pyridin-2-ylmethylbutyramide (27): ¹H NMR (CDCl₃, 300 MHz) δ = 2.23–2.27 (m, 6H), 2.45 (t, 1H, $J = 7.4$ Hz), 2.55 (t, 1H, $J = 7.3$ Hz), 2.83 (t, 1H, $J =$ 6.4 Hz), 2.91 (t, 2H, $J = 6.4$ Hz), 3.37 (t, 1H, $J = 6.4$ Hz), 3.43 (t, 1H, J = 6.4 Hz), 3.53−3.61 (m, 2H), 4.74−4.80 (m, 2H), 7.13−7.23 (m, 1H), 7.25−7.38 (m, 1H), 7.42−7.48 (m, 2H), 7.52−7.58 (m, 1H), 7.61−7.73 (m, 1H), 7.98−8.04 (m, 2H), 8.49−8.59 (m, 1H); 13C NMR (CDCl₃, 100 MHz) δ = 27.0, 27.5, 33.7, 33.9, 44.6, 45.6, 45.8, 46.5, 51.8, 53.8, 56.9, 57.8, 120.8, 122.2, 122.6, 128.1, 128.6, 133.1, 136.8, 137.1, 149.2, 149.8, 157.2, 158.0, 172.2, 172.5, 199.2; MS (CI, NH₃) $m/z = 340.2$ (100) [(M + 1)⁺]; HRMS (CI, NH₃) m/z [(M + $[1)^+$] calcd for $C_{20}H_{25}N_3O_2$ 340.20250, found 340.20260.

4-Cyano-N-(2-dimethylaminoethyl)-N-(2-iminobut-3-enyl) benzamide (28). Amine 9a was acylated according to the general procedure b, and benzamide 28 was obtained as a yellow gum (0.22 g, 0.7 mmol, 40%) after purification with column chromatography (silica gel, DCM/MeOH/Et₃N 99:1:0.1 \rightarrow DCM/MeOH/Et₃N 97:3:0.1).

4-Cyano-N-(2-dimethylaminoethyl)-N-(2-iminobut-3-enyl) benzamide (28): ¹H NMR (CDCl₃, 300 MHz) δ = 1.96–2.32 (m, 6H), 2.32−2.62 (m, 2H), 3.30−3.67 (m, 2H), 4.56−4.97 (m, 2H), 7.07−7.46 (m, 1H), 7.22 (ddd, 1H), 7.55−7.76 (m, 5H), 8.58 (m_c, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ = 45.7, 113.4, 118.3, 122.7, 127.8, 132.3, 136.9, 141.0, 170.4; MS (CI, NH₃) $m/z = 309.2$ (100) $[(M + 1)^+]$; HRMS (APCI, MeOH) m/z $[(M + 1)^+]$ calcd for $C_{18}H_{20}N_4O$ 309.17154, found 309.17090.

N-(2-Dimethylaminoethyl)-4-methoxy-N-pyridin-2-ylmethylbenzamide (29). Amine 9a was acylated according to the general procedure b and benzamide 29 was obtained as a yellow gum (0.31 g, 1.0 mmol, 37%) after purification with column chromatography (silica gel, DCM/MeOH/Et₃N 99:1:0.1 \rightarrow DCM/MeOH/Et₃N 97:3:0.1).

N-(2-Dimethylaminoethyl)-4-methoxy-N-pyridin-2-ylmethylbenzamide (29): ¹H NMR (CDCl₃, 300 MHz) δ = 1.97–2.34 (m, 6H), 2.34−2.65 (m, 2H), 3.36−3.67 (m, 2H), 3.81 (s, 3H), 4.64−4.97 (m, 2H), 6.89 (m_c, 2H), 7.19 (ddd, 1H), 7.22–7.47 (m, 3H), 7.68 (dt, 1H), 8.56 (m_o, 1H); ¹³C NMR (CDCl₃, 100 MHz) $\delta = 45.7$, 55.5, 113.7, 122.5, 128.8, 136.9, 160.7, 172.2; MS (pos ESI) m/z = 314.2 (100) $[(M + 1)^+]$; HRMS (pos ESI, MeCN) m/z $[(M + 1)^+]$ calcd for $C_{18}H_{23}N_3O_2$ 314.18685, found 314.18630.

■ ASSOCIATED CONTENT

S Supporting Information

 1 H and 13 C NMR spectra and theoretical calculation data. This material is available free of charge via the Internet at http:// pubs.acs.org.

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Notes

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■ REFERENCES

(1) Kociénski, P. J. Protecting Groups, 3rd ed.; Georg Thieme: Stuttgart, 2003.

(2) Greene, T. W.; Wuts, P. G. M. Protective Groups in Organic Synthesis, 3rd ed.; Wiley-Interscience: New York, 1999.

(3) Bröhmer, M. C.; Mundinger, S.; Bräse, S.; Bannwarth, W. Angew. Chem., Int. Ed. 2011, 50.

(4) Kramer, R. A.; Brö hmer, M. C.; Forkel, N. V.; Bannwarth, W. Eur. J. Org. Chem. 2009, 4273−4283.

(5) Bröhmer, M. C.; Bannwarth, W. Eur. J. Org. Chem. 2008, 4412− 4415; Synfacts 2008, 11, 1226−1226.

(6) Inoue, K.; Sakai, K. Tetrahedron Lett. 1977, 46, 4063−4066.

(7) Patterson, I.; Yeung, K.-S.; Ward, R. A.; Smith, J. D.; Cumming, J. G.; Lamboley, S. Tetrahedron Lett. 1995, 51, 9467−9486.

(8) Nambu, M.; White, J. D. Chem. Commun. 1996, 1619−1620.

(9) This compound was obtained from Prof. Xile Hu, University of Lausanne.

(10) Csok, Z.; Vechorkin, O.; Harkins, S. B.; Scopelliti, R.; Hu, X. L. J. Am. Chem. Soc. 2008, 130, 8156−8157.

(11) Knorr, R.; Trzeciak, A.; Bannwarth, W.; Gillessen, D. Tetrahedron Lett. 1989, 30, 1927−1930.

(12) Bandyopadhyay, P.; Jha, S.; Imran Ali, S. K. J. Agric. Food Chem. 2009, 57, 9780−9786.

(13) Sundaravel, K.; Suresh, E.; Saminathanc, K.; Palaniandavar, M. Dalton Trans. 2011, 40, 8092−8107.

(14) Bebout, D. C.; Lai, W.; Stamps, S. M.; Berry, S. M.; Butcher, R. J. Polyhedron 2008, 27, 1591−1600.

(15) Bhattacharya, S.; Snehalatha, K.; Kumar, V. P. J. Org. Chem. 2003, 68, 2741−2747.

(16) Brunner, H.; Niemetz, M. Monatsh. Chem. 2002, 133, 115−126. (17) Blank, B.; Madalska, M.; Kempe, R. Adv. Synth. Catal. 2008, 350, 749−758.

(18) Foxon, S. P.; Walter, O.; Schindler, S. Eur. J. Inorg. Chem. 2002, 111−121.

(19) Magnus, G.; Levine, R. J. Am. Chem. Soc. 1956, 78, 4127−4130.

(20) Bangov, P.; Radeglia, R. Org. Magn. Reson. 1983, 21, 443−449.

(21) Kumar, P.; Kalita, A.; Mondal, B. Dalton Trans. 2011, 40, 8656− 8663.

(22) Adams, H.; Bailey, N. A.; de Barbarin, C. O. R.; Fenton, D. E.; He, Q.-Y. J. Chem. Soc., Dalton Trans. 1995, 2323-2331.

(23) Leaver, S.; Palaniandavar, A. M.; Kilner, C.; Halcrow, M. Dalton Trans. 2003, 22, 4224−4225.

(24) Struthers, H.; Spingler, B.; Mindt, T. L.; Schibli, R. Chem.-Eur. J. 2008, 14, 6173−6183.

(25) Balaraman, E.; Gnanaprakasam, B.; Shimon, L. J. W.; Milstein, D. J. Am. Chem. Soc. 2010, 132, 16756-16758.

(26) Heirtzler, F. R.; Neuburger, M.; Zehnder, M.; Constable, E. C. Liebigs Ann. 1997, 297−301.

(27) Katritzky, A. R.; Scriven, E. F. V.; Majumder, S.; Akhmedova, R. G.; Vakulenko, A. V.; Akhmedov, N. G.; Murugan, R.; Abboud, K. A. Org. Biomol. Chem. 2005, 3, 538−541.

(28) Liu, M.-C.; Lin, T.-S.; Sartorelli, A. C. J. Med. Chem. 1992, 35, 3672−3677.

(29) Swanson, D. M.; Shah, C. R.; Lord, B.; Morton, K.; Dvorak, L. K.; Mazur, C.; Apodaca, R.; Xiao, Wei; Boggs, J. D.; Feinstein, M.; Wilson, S. J.; Barbier, A. J.; Bonaventure, P.; Lovenberg, T. W.; Carruthers, N. I. Eur. J. Med. Chem. 2009, 44, 4413−4425.

(30) Luitjes, H.; Schakel, M.; Klumpp, G. W. Synth. Commun. 1994, 24, 2257−2262.